

# **SAFETY DATA SHEET**

# 1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

1.1 Product identifier

Product name ARGON, COMPRESSED

Synonym(s) ARGON • Ar • PRODUCT CODES: A1000 • SHIELDING GAS FOR WELDING

1.2 Uses and uses advised against

Use(s) ANALYTICAL CHEMISTRY ● INDUSTRIAL APPLICATIONS

1.3 Details of the supplier of the product

Supplier name Industrial Gases New Zealand Ltd t/a Eziswap Gas

Address 6 and 10 Canaveral Drive, Rosedale, Auckland, NEW ZEALAND

**Telephone** +64 9 444 0357 **Fax** +64 9 444 3509

Emailsales@eziswapgas.co.nzWebsitehttp://www.eziswapgas.co.nz

1.4 Emergency telephone number(s)

Emergency 111 (NZ only)

## 2. HAZARDS IDENTIFICATION

# 2.1 Classification of the substance or mixture

NOT CLASSIFIED AS HAZARDOUS ACCORDING TO HAZARDOUS SUBSTANCES [CLASSIFICATION] REGULATIONS 2001

HSNO classification(s)

Compressed gases Contains gas under pressure; may explode if heated.

2.2 Label elements

Signal word WARNING

Pictogram(s)



Hazard statement(s)

H280 Contains gas under pressure; may explode if heated.

Prevention statement(s)

P103 Read label before use.

Response statement(s)

None allocated.

Storage statement(s)

P410 + P403 Protect from sunlight. Store in a well-ventilated place.

Disposal statement(s)

None allocated.

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#### 2.3 Other hazards

Asphyxiant. Effects are proportional to oxygen displacement.

## 3. COMPOSITION/ INFORMATION ON INGREDIENTS

#### 3.1 Substances / Mixtures

Ingredient	CAS Number	EC Number	Content (v/v)
ARGON	7440-37-1	231-147-0	>99.9%

# 4. FIRST AID MEASURES

## 4.1 Description of first aid measures

Eye None required.

Inhalation If inhaled, remove from contaminated area. To protect rescuer, use an Air-line respirator or Self Contained

Breathing Apparatus (SCBA). Apply artificial respiration if not breathing. Give oxygen if available.

Skin None required.

**Ingestion** Ingestion is not considered a potential route of exposure.

First aid facilities None allocated.

## 4.2 Most important symptoms and effects, both acute and delayed

In high concentrations may cause asphyxiation. Symptoms may include loss of mobility / consciousness. Victim may not be aware of asphyxiation.

#### 4.3 Immediate medical attention and special treatment needed

Treat symptomatically.

# 5. FIRE FIGHTING MEASURES

#### 5.1 Extinguishing media

Use water fog to cool containers from protected area.

## 5.2 Special hazards arising from the substance or mixture

Non flammable.

## 5.3 Advice for firefighters

Temperatures in a fire may cause cylinders to rupture. Cool cylinders or containers exposed to fire by applying water from a protected location. Remove cool cylinders from the path of the fire. Evacuate the area if unable to keep cylinders cool. Do not approach cylinders or containers suspected of being hot.

#### 5.4 Hazchem code

2T

2 Fine Water Spray.

T Wear full fire kit and breathing apparatus. Dilute spill and run-off.

# 6. ACCIDENTAL RELEASE MEASURES

# 6.1 Personal precautions, protective equipment and emergency procedures

If the cylinder is leaking, evacuate area of personnel. Inform manufacturer/supplier of leak. Use Personal Protective Equipment (PPE) as detailed in Section 8 of the SDS.

#### 6.2 Environmental precautions

Prevent from entering sewers, basements and workpits, or any place where its accumulation can be dangerous.

#### 6.3 Methods of cleaning up

Carefully move material to a well ventilated remote area, then allow to discharge if safe to do so. Do not attempt to repair leaking valve or cylinder safety devices.

# 6.4 Reference to other sections

See Sections 8 and 13 for exposure controls and disposal.

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# 7. HANDLING AND STORAGE

#### 7.1 Precautions for safe handling

Use of safe work practices are recommended to avoid inhalation. Do not drag, drop, slide or roll cylinders. The uncontrolled release of a gas under pressure may cause physical harm. Use a suitable hand truck for cylinder movement.

## 7.2 Conditions for safe storage, including any incompatibilities

Cylinders should be stored below 45°C in a secure area, upright and restrained to prevent cylinders from falling. Cylinders should also be stored in a dry, well ventilated area constructed of non-combustible material with firm level floor (preferably concrete), away from areas of heavy traffic and emergency exits.

#### 7.3 Specific end use(s)

No information provided.

# 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

## 8.1 Control parameters

#### **Exposure standards**

Ingredient Refere	Reference		<b>VA</b>	STEL	
Ingredient	Kelefelice	ppm	mg/m³	ppm	mg/m³
Argon	WES (NZ)	Asphyxiant			

# **Biological limits**

No biological limit values have been entered for this product.

#### 8.2 Exposure controls

Engineering controls Provide suitable ventilation to minimise or eliminate exposure. Confined areas (e.g. tanks) should be

adequately ventilated or gas tested.

PPE

**Eye / Face** Wear safety glasses. **Hands** Wear leather gloves.

**Body** Wear coveralls and safety boots.

Respiratory Where an inhalation risk exists, wear Self Contained Breathing Apparatus (SCBA) or an Air-line respirator.









# 9. PHYSICAL AND CHEMICAL PROPERTIES

## 9.1 Information on basic physical and chemical properties

Appearance COLOURLESS GAS
Odour ODOURLESS
Flammability NON FLAMMABLE
Flash point NOT RELEVANT
Boiling point -185.9°C

**NOT AVAILABLE Melting point NOT APPLICABLE Evaporation rate** рΗ **NOT APPLICABLE** Vapour density **NOT AVAILABLE** Specific gravity NOT APPLICABLE Solubility (water) 0.0337 cm<sup>3</sup>/cm<sup>3</sup> Vapour pressure **NOT AVAILABLE** Upper explosion limit NOT RELEVANT Lower explosion limit **NOT RELEVANT NOT AVAILABLE** Partition coefficient **NOT AVAILABLE** Autoignition temperature **NOT AVAILABLE Decomposition temperature** 

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## 9.1 Information on basic physical and chemical properties

Viscosity

Explosive properties

Oxidising properties

Odour threshold

NOT AVAILABLE

NOT AVAILABLE

NOT AVAILABLE

9.2 Other information

% Volatiles 100 %

**Cylinder pressure (when full)** 13000 - 30000 kPa @ 15°C **Density** 1.667 kg/m³ @ 15°C

## 10. STABILITY AND REACTIVITY

#### 10.1 Reactivity

Carefully review all information provided in sections 10.2 to 10.6.

#### 10.2 Chemical stability

Stable under recommended conditions of storage.

#### 10.3 Possibility of hazardous reactions

Polymerization will not occur.

# 10.4 Conditions to avoid

Avoid shock, friction, heavy impact, heat, sparks, open flames and other ignition sources.

#### 10.5 Incompatible materials

Compatible with most commonly used materials. Hazardous by-products may be produced when this gas/gas mixture is used in welding, cutting and associated processes.

#### 10.6 Hazardous decomposition products

This material will not decompose to form hazardous products other than that already present.

# 11. TOXICOLOGICAL INFORMATION

# 11.1 Information on toxicological effects

Acute toxicity Based on available data, the classification criteria are not met.

Skin Not classified as a skin irritant.

Eye Not classified as an eye irritant.

**Sensitisation** Not classified as causing skin or respiratory sensitisation.

MutagenicityNot classified as a mutagen.CarcinogenicityNot classified as a carcinogen.ReproductiveNot classified as a reproductive toxin.

STOT - single Asphyxiant. Effects are proportional to oxygen displacement. Over exposure may result in dizziness,

**exposure** drowsiness, weakness, fatigue, breathing difficulties and unconsciousness.

STOT - repeated

exposure

Not classified as causing organ damage from repeated exposure.

**Aspiration** Not classified as causing aspiration.

## 12. ECOLOGICAL INFORMATION

# 12.1 Toxicity

No information provided.

# 12.2 Persistence and degradability

No information provided.

#### 12.3 Bioaccumulative potential

No information provided.

#### 12.4 Mobility in soil

No information provided.

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## 12.5 Other adverse effects

Product is not harmful to the environment. ATMOSPHERE: Argon is an inert gas and will not contribute to ozone depletion or global warming. Argon is a natural component of the atmosphere (0.93% v/v). SOIL/WATER: If released to soil or water argon will quickly evaporate to the atmosphere. Not toxic to plants or animals except at extremely high (asphyxiating) levels. Fume from fabrication processes which use this gas/gas mixture may be harmful to the environment.

# 13. DISPOSAL CONSIDERATIONS

#### 13.1 Waste treatment methods

Waste disposal Cylinders should be returned to the manufacturer or supplier for disposal of contents.

**Legislation** Dispose of in accordance with relevant local legislation.

# 14. TRANSPORT INFORMATION

CLASSIFIED AS A DANGEROUS GOOD ACCORDING TO LAND TRANSPORT RULE: DANGEROUS GOODS 2005; NZS 5433:2012, UN, IMDG OR IATA



	LAND TRANSPORT (NZS 5433)	SEA TRANSPORT (IMDG / IMO)	AIR TRANSPORT (IATA / ICAO)
14.1 UN Number	1006	1006	1006
14.2 Proper Shipping Name	ARGON, COMPRESSED	ARGON, COMPRESSED	ARGON, COMPRESSED
14.3 Transport hazard class	2.2	2.2	2.2
14.4 Packing Group	None allocated.	None allocated.	None allocated.

## 14.5 Environmental hazards

No information provided.

#### 14.6 Special precautions for user

Hazchem code 2T EMS F-C, S-V

Other information Ensure cylinder is separated from driver and that outlet of relief device is not obstructed.

# 15. REGULATORY INFORMATION

## 15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

Approval code HSR001017
Group standard Argon

Inventory listing(s) NEW ZEALAND: NZIoC (New Zealand Inventory of Chemicals)

All components are listed on the NZIoC inventory, or are exempt.

## **16. OTHER INFORMATION**

Additional information

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#### ARGON, COMPRESSED PRODUCT NAME

When using this gas/gas mixture for welding, cutting and associated processes, additional hazards may be generated by the process such as radiation, noise and fume. Risk assessments should be made for each activity to identify and quantify the individual hazards involved. Refer to the relevant Safety Data Sheets for the welding consumables being used or, if available, the materials being welded. This product is used in scientific analysis as a purge gas in emission spectroscopy and as a carrier gas in gas chromatography. Argon mixtures are used as x-ray counting gases. Argon is used extensively in the incandescent lamp industry for the filling of light bulbs. It is used in arc welding as an inert gas shield to prevent oxidation of the metals being welded. It is also used with other rare gases in the filling of special bulbs and display tubes to obtain special colour effects in the neon-type bulbs. Many geiger-counting tubes contain argon or argon mixed with organic vapours and gases.

APPLICATION METHOD: Gas regulator of suitable pressure and flow rating fitted to cylinder or manifold with low pressure gas distribution to equipment.

#### PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:

The recommendation for protective equipment contained within this report is provided as a guide only. Factors such as form of product, method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is made.

#### HEALTH EFFECTS FROM EXPOSURE:

It should be noted that the effects from exposure to this product will depend on several factors including: form of product; frequency and duration of use; quantity used; effectiveness of control measures; protective equipment used and method of application. Given that it is impractical to prepare a report which would encompass all possible scenarios, it is anticipated that users will assess the risks and apply control methods where appropriate.

Abbreviations	ACGIH	American Conference of Governmental Industrial Hygienists
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CAS# Chemical Abstract Service number - used to uniquely identify chemical compounds

CCID Chemical Classification and Information Database (HSNO)

Central Nervous System CNS

EC No. EC No - European Community Number

**EMS** Emergency Schedules (Emergency Procedures for Ships Carrying Dangerous

Goods)

**EPA** Environmental Protection Authority [New Zealand]

GHS Globally Harmonized System

Hazardous Substances and New Organisms **HSNO** International Agency for Research on Cancer IARC

LC50 Lethal Concentration, 50% / Median Lethal Concentration

LD50 Lethal Dose, 50% / Median Lethal Dose

mg/m<sup>3</sup> Milligrams per Cubic Metre OEL Occupational Exposure Limit

relates to hydrogen ion concentration using a scale of 0 (high acidic) to 14 (highly рΗ

Parts Per Million ppm

Short-Term Exposure Limit STEL

STOT-RE Specific target organ toxicity (repeated exposure) STOT-SE Specific target organ toxicity (single exposure)

Threshold Limit Value TLV Time Weighted Average **TWA** 

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